

Approved	by:	••••	••••	•••	•••	••	• •	••	•	••

# Dean

Date.			
Date.	• • • • • • • •	 •••••	••••

# SOFIA UNIVERSITY "ST. KLIMENT OHRIDSKI"

••••

(code and name)

Biochemistry

Lecturer: Prof. Dr. Mariela Odjakova, Prof. Dr. Svetla Petrova

Teaching assistant: Assist, Prof. Dr. Lyuben Zagorchev

Academic work	Туре	Acad. hours
In-class work	Lectures	45
	Seminars	30
	Practical classes (teaching assistance)	30
Total in-class work		105
Out-of-class work	Topical writing /Course paper	15
	Presentation	15
	Scientific essay	
	Course project	
	Field trip	
	Independent literature research	
	Student teaching	40
Total out-of-class work		70
TOTAL ACADEMIC WORK		175
ECTS credits in-class work		4.2
ECTS credits out-of-class work		2.8
TOTAL ECTS CREDITS		7

N₂	Grade components <sup>1</sup>	% of the grade
1.	Workshops {search of information and group discussions of presentations and topical writings)	
2.	Participation in topical discussions during the classes	5
3.	Demonstrations in class	
4.	Field trip attendance	
5.	Portfolio	
6.	Quizzes throughout the semester	15
7.	Case studies	15
8.	Homeworks/mid-term test	15
9.		
10.		
11.		
12.	Final exam	50

## **Outline of the course:**

The Biochemistry course covers the structural, functional and informational aspects of biochemical processes. The major classes of biomolecules and central metabolic pathways are considered with accent on degradation and biosynthesis of biochemically important molecules (proteins, carbohydrates, lipids and nucleic acids). Particular attention is put on the organisation and regulation of the metabolic processes. The basic principles of expression and transfer of genetic information, which explains the mechanisms of biochemical identity of organisms are considered through replication, transcription and protein synthesis in prokaryotic and eukaryotic organisms. The importance of the biochemical approach for the development of other fundamental (pharmacology, human physiology, genetics, clinical laboratory etc.) and applied (drug design) biosciences is demonstrated with appropriate examples. Laboratory exercises illustrate and expand the lectures and establish lab skills and habits, needed for the future professional realization of students.

## **Preliminary requirements:**

Students must have successfuly completed the courses in organic chemistry, biology and physiology.

#### **Key competences acquired:**

Students acquire basic knowledge on the fundamental biochemical processes, structure and function of biomolecules and key metabolic pathways and general lab skills, needed for successful application of the knowledge acquired in different fields of biochemistry, physiology and pharmacology.

Depending on the course specificity and on the requirements of the teacher, other types of activity can be added or the unnecessary ones can be removed.

# Lessons plan

No	Topic:	Acad. hours
	Lectures	
1	Subject of biochemistry. Molecular organization of living	1
	matter. Basic principles. Water — structure, properties and	
	functions. Buffer systems in living organisms.	
2	<b>Proteins.</b> Amino acids, peptides and polypeptides. Levels of	5
	organization of the protein molecule. Domain structure.	
	Structure-function relationship. Protein folding, native	
	conformation. Denaturation and renaturation, chaperones.	
	Protein modifications. Principles of protein purification and analysis. Oxygen-binding proteins — hemoglobin. Proteins of	
	the interstitial matrix: collagens, proteoglycans, glycoproteins,	
	adhesive proteins.	
3	<b>Biocatalysts.</b> Enzymes and ribozymes. Cofactors. Active site –	4
	structure and properties. Enzymes classification. Mechanisms	•
	of the enzymatic catalysis. Enzyme-substrate complex. Enzyme	
	kinetics – Michaelis-Menten model; Km, Vmax, kcat, kcat/Km	
	criteria. Regulation of enzyme activity, impact of temperature	
	and pH. Inhibition, irreversible reversible inhibition. Allosteric	
	enzymes. Analogues of the transit state, catalytic antibodies	
	(abzymes). Clinical importance of enzymes.	
4.	Catalytic strategies. Basic principles of catalysis. Catalysis of	2
	hardly proceeding reactions: proteases – serine proteases	
	(chymotripsin, tripsin, elastase), aspartate-containing (pepsin,	
	HIV-protease), metal-containing. Drugs – protease inhibitors.	
5.	Catalytic mechanism of cyclooxygenase – drugs - inhibitors	
٥.	<b>Regulatory strategies.</b> Allosteric control, product inhibition. Symmetry and sequential models of allosterism. Cooperative	2
	effect. Hemoglobin, oxygen binding, effect of 2,3-	2
	bisphosphoglycerate. Bor effect. Isoenzymes – lactat	
	dehydrogenase. Regulation of the enzyme activity through	
	covalent modifications. Regulation of the enzyme activity	
	through proteolysis – digestive and blood clotting enzymes.	
6.	Metabolic strategies Metabolism – general characteristics,	1
	metabolic chains, reversible and irreversible reactions, key	
	metabolites. Compartmentalization. Entropy and laws of	
	thermodinamics. Exergonic and endergonic reactions.	
	Biological oxidation. High energy compounds. Anabolitic and	
	catabolitic, aerobic and anaerobic processes. Role of ATP,	
	NAD and NADP. Energetical status of the cell. Basic	
7	principles of metabolic regulation.	1
7.	Carbohydrates. Monosaccharides, disaccharides and	1
	polysaccharides. Glycoproteins – O and N-linked.	
	Proteoglycans. Biologically active glycosides.	

**Carbohydrate metabolism.** Glycolysis. ATP synthesis at the 6 substrate level. Glycolysis and metabolism of erytrocytes. Anaerobic glycolysis, lactic acidosis and Cori cycle. Cancer and glycolysis. Lactose, fructose and ethanol metabolism. Galactose, galactosemia. Glycogen metabolism, regulation, metabolic disorders. glycogenosis. Gluconeogenesis. Regulation of glycolysis and gluconeogenesis. Substrate cycles and heat production. Blood sugar control, hypoglycemia. Pentose-phosphate pathway. Citric acid cycle. Piruvate dehydrogenase complex. Anaplerotic reactions. Regulation. Oxidative phosphorylation and bioenergetics. Energy accumulation in the cell. High energy compounds as mediators between exergonic and endergonic reactions. Importance of ATP for the energetic turnover in the cell. Redox systems with biological importance. Substrate chains of the biological oxidation. Molecular design of the electron transport chains. Electron transport chain linked dehydrogenases. Mechanisms of oxidative phosphorylation. ATP synthase. Shuttle mechanisms for electron and metabolites transport. Regulation of cellular respiration. Oxidative phosphorylation inhibitors. Mitochondria and apoptosis. Electron transport chains in the endoplasmic reticulum, generation and scavenging of superoxide, hydrogen peroxide and free radicals. **Lipids.** Fatty acids, phospholipids, sphingolipids, glycolipids, 2 10. steroids, lipoproteins. Biological membranes, properties. Membrane proteins. Transport accross membrane channels and pumps. Liposomes. Application for drug therapy. **Biosignaling.** General features of signal transduction. G-protein 2 11. coupled receptors and second messengers. Receptor tyrosine kinases. Receptor guanylyl cyclases. Regulation of the cell cycle by protein kinases. Oncogenes, tumor suppressor genes and programmed cell death. Development of protein kinases inhibitors for cancer treatment. 12. Degradation of Lipid metabolism. triacylglycerols 3 Chilomicrones. Degradation of triacylglycerols in fatty tissues. Fatty acids oxidation. Peroxisomal oxidation. Ketogenesis, regulation of lipid metabolism. Metabolism of phospholipids, sphingolipids and eicosanoids. Lipid transport. Lipoprotein complexes, types, formation and biological importance. Receptor-mediated endocytosis of lipoproteins. Metabolism of cholesterol. Bile acids, steroid hormones, vitamin D. Metabolism of arachidonic acid. Regulation of lipid metabolism. Drugs, inhibitors of lipid biosynthesis. 13. **Amino** acids metabolism. Oxidative deamination, 2 transamination. decarboxylation. Biogenic amines. Detoxification of ammonia. Urea cycle. Degradation of carbon

	skeleton – glucogenic and ketogenic amino acids. Role of folio acid in single carbon radicals transfer. Biosynthesis of amino acids, essential and non-essential amino acids. Amino acids as biosynthetic precurssors. Porphyrins and bile pigments. Drugs,	
	targeting amino acids metabolism.	
14.	<b>Nucleotidic metabolism.</b> Synthesis of purine and pirimidine nucleotides. Degradation of nucleotides. Drugs, targeting nucleotidic metabolism.	
15.	<b>Nucleic acids.</b> Types, structure, level of organization, functions.	2
16.	<b>DNA metabolism.</b> Biosynthesis of DNA (replication). Enzymatic machinery. Mechanism of replication. Inhibitors of replication.	
17.	<b>RNA metabolism.</b> Biosynthesis of RNA (transcription). Inhibitors of transcription.	2
18.	Protein metabolism. Biosynthesis of proteins (translation).	
	Major stages. Posttranslational modifications. Degradation of proteins. Inhibitors of translation.	
19.	Integration of metabolism. Strategies for metabolic regulation.	
	Interrelation between metabolism of carbohydrates,, fats and	
	amino acids. Metabolism at molecular, cellular and organ level.  Seminars/Practical exercises	
1	Calculating concentrations problems. Buffer solutions. Amino	6
	acids properties. Titration curves. Isoelectric point calculation.	
2	Isolation of protein by fractional precipitation, methods for desalting of fractions (dialysis, gel filtration, ultrafiltration).	6
3	Protein characterization. Determination of the absorbance spectrum. Methods for quantitative characterization. Standard curves.	6
4.	Protein characterization. Protein fractionation with electrophoresis in polyacrylamide gel (PAGE) in denaturing and native conditions. Molecular weight determination - SDS PAGE, gel filtration, FPLC, HPLC. Sediment analysis.	8
5.	Carbohydrates. Standard curves — DNS and Somogyi-Nelson methods.	4
6.	Enzyme kinetics. Initial rate determination. Effect of enzyme concentration on the reaction rate. Units for enzyme acitivity.	6
7.	Effect of subtrate concentration on the reaction rate. Michaelis-Menten constant (Km) determination. Substrate inhibition.	6
	Kinetics of product inhibition.	
8.	Effect of inhibitors on the reaction rate. Determination of ID50	6
0	of drugs.	6
9.	Genomic and plasmidic DNA isolation from E. Coli. Spectrophotometrical and electrophoretical characterization of	6
10.	nucleic acids. Lipids. Quantitation of cholesterol.	6
10.	Diplos. Quantumion of environment.	0

# Topics Covered on the Final Exam

N₂	Topic
1	Chemical bonds. Water – structure and functions. Properties. pH. Buffers.
2	Proteins. Amino acids, peptides and polypeptides. Levels of organization of the
_	protein molecule.
3	Link between structure and functions. Globular and fibrillar proteins. Domain structure
	Structure-function relationship. Protein folding, native conformation. Denaturation
	and renaturation, chaperones.
4.	Oxygene-transfer proteins. Hemoglobin – oxygen binding, effect of 2,3
	bisphosphoglycerate. Bor effect. Protein modifications. Principles of protein
	purification and analysis.
5.	Biocatalysts. Enzymes and ribozymes. Cofactors. Active site - structure an
	properties. Enzymes classification.
6.	Mechanisms of the enzymatic catalysis. Enzyme-substrate complex. Enzyme kinetic
	– Michaelis-Menten model; Km, Vmax, kcat, kcat/Km criteria.
7.	Regulation of enzyme activity, impact of temperature and pH.
8.	Inhibition, irreversible and reversible (competitive and non-competitive) inhibition
	Allosteric enzymes. Allosteric control, product inhibition. Cooperative effect
	Analogues of the transit state, catalytic antibodies (abzymes).
9.	Catalytic strategies. General catalytic mechanisms. Serine proteinases as example for catal
10.	of hardly proceeding reactions and catalytic mechanims of lyzosyme.  Metabolic strategy. Metabolism – general characteristics, metabolic chains, reversi
10.	and irreversible reactions, key metabolites. Compartmentalization. Anabolitic
	catabolitic, aerobic and anaerobic processes. Entropy and laws of thermodinam
	Exergonic and endergonic reactions.
11.	Energetical status of the cell. Role of ATP. High energy compounds. Role of NA
	NADP and FAD. Basic principles of metabolic regulation.
12.	Carbohydrates. Monosaccharides, disaccharides and polysaccharides.
	Glycoproteins – O and N-linked.
13.	Carbohydrate metabolism. Glycolysis. ATP synthesis at the substrate level. Glycoly
	and metabolism of erytrocytes. Anaerobic glycolysis, lactic acidosis and Cori cy
	Cancer and glycolysis.
14.	Glycogen metabolism, regulation.
15.	Gluconeogenesis. Regulation of glycolysis and gluconeogenesis. Substrate cycles
4 -	heat production.
16.	Pentose phosphate pathway.
17.	Citric acid cycle. Piruvate dehydrogenase complex. Anaplerotic reactions. Regulation
18.	Oxidative phosphorilation and bioenergetics. Electron transport chains, proton pun
19.	transmembrane proton transport – Q cycle.  Mitschondrie and aportosis Electron transport in the andonlasmic raticulum generation.
17.	Mitochondria and apoptosis. Electron transport in the endoplasmic reticulum, generation scavenging of superoxide, hydrogen peroxide and free radicals.
20.	Lipids. Fatty acids, phospholipids, sphingolipids, glycolipids, steroids, lipoproteins.
20.	Biological membranes.
21.	Membrane proteins. Transport accross membrane channels and pumps. Liposomes.
	Application for drug therapy.
<u> </u>	

- 22. Biosignaling. General features of signal transduction. G-protein coupled receptors and second messengers. Receptor tyrosine kinases. Receptor guanylyl cyclases.
- 23. Regulation of the cell cycle by protein kinases. Oncogenes, tumor suppressor genes and programmed cell death. Development of protein kinases inhibitors for cancer treatment.
- 24. Degradation of triacylglycerols. Chilomicrones. Degradation of triacylglycerols in fatty tissues. Fatty acids oxidation. Peroxisomal oxidation. Ketogenesis, regulation of lipid metabolism.
- 25. Fatty acids biosynthesis. Lipid metabolism regulation. Metabolism of phospholipids sphingolipids and eicosanoids.
- 26. Lipid transport. Lipoprotein complexes, types, formation and biological importance. Receptor-mediated endocytosis of lipoproteins. Metabolism of cholesterol. Bile acids, steroid hormones, vitamin D.
- 27. Metabolism of arachidonic acid. Regulation of lipid metabolism. Drugs, inhibitors of lipid biosynthesis.
- 28. Amino acids metabolism. Oxidative deamination, transamination, decarboxylation Biogenic amines. Detoxification of ammonia. Urea cycle.
- 29. Degradation of carbon skeleton glucogenic and ketogenic amino acids. Role of folic acid in single carbon radicals transfer.
- 30. Biosynthesis of amino acids, essential and non-essential amino acids. Amino acids as biosynthetic precurssors. Porphyrins and bile pigments. Drugs, targeting amino acids metabolism.
- 31. Nucleotidic metabolism. Synthesis of purine and pirimidine nucleotides Degradation of nucleotides. Drugs, targeting nucleotidic metabolism.
- 32. Nucleic acids. Types, structure, level of organization, functions.
- 33. DNA metabolism. Biosynthesis of DNA (replication). Enzymatic machinery. Mechanism of replication. Inhibitors of replication.
- 34. RNA metabolism. Biosynthesis of RNA (transcription). Inhibitors of transcription.
- 35. Protein metabolism. Biosynthesis of proteins (translation). Major stages. Posttranslational modifications. Degradation of proteins. Inhibitors of translation.
- 36. Integration of metabolism. Strategies for metabolic regulation. Interrelation between metabolism of carbohydrates, fats and amino acids. Metabolism at molecular, cellular and organ level.

## **Bibliography**

## Main sources:

J.M. Berg., L. Stryer., J.C. Tymoczko et al., 2011, Biochemistry. W.H.Freeman & Co.

D. Voet & J. Voet, 2010, **Biochemistry**. J. Wiley & Sons Inc.

A. Lehninger, D.Nelson., M.Cox., 2008, **Lehninger Principles of Biochemistry**. Worth Publishers C.K. Mathews., K.E. van Holde., K.G. Ahern et al., 2012, **Biochemistry**. Benjamin.Cummings Publ

L. A. Moran, R.A. Horton, G.Scrimgeour and M. Perry., 2011, **Principles of Biochemistry**. Prentice Hall Publ.